

Palladium(II)-Catalyzed *meta*-C–H Olefination: Constructing Multisubstituted Arenes through Homo-Diolefinatation and Sequential Hetero-Diolefinatation**

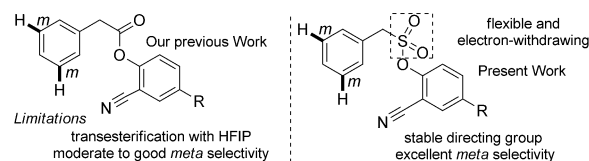
Milan Bera, Arun Maji, Santosh K. Sahoo, and Debabrata Maiti*

Abstract: Divinylbenzene derivatives represent an important class of molecular building blocks in organic chemistry and materials science. Reported herein is the palladium-catalyzed synthesis of divinylbenzenes by *meta*-C–H olefination of sulfone-based arenes. Successful sequential olefinations in a position-selective manner provided a novel route for the synthesis of hetero-dialkenylated products, which are difficult to access using conventional methods. Additionally, 1,3,5-trialkenylated compounds can be generated upon successful removal of the directing group.

Transition-metal-catalyzed functionalization of arene C–H bonds has been extensively investigated over the last two decades by using a variety of directing groups (DGs).^[1] In the C–H activation realm, directing groups are almost universally *ortho*-selective.^[2] Compared to *ortho*-functionalization, the related *meta*-selective reactions pose a significant challenge. In this regard, examples of *meta*-selective functionalization have only recently been reported and are based on steric and electronic bias.^[3] An alternate template-based approach, initiated by the group of Yu^[4] and subsequently extended by Tan and co-workers^[5] as well as our group,^[6] demonstrated *meta*-selective functionalization in the presence of a suitable directing group. These *meta*-selective C–H functionalizations are usually independent of the substitution pattern and electronic nature of the substrate. Importantly, activation of *meta* C–H bonds hinges upon the combined use of a weakly coordinating nitrile-based template with a protected amino-acid ligand.

During our previous work on palladium-catalyzed *meta*-C–H olefination with arylacetic acid derived scaffolds,^[6] a number of limitations were encountered: 1) unsuccessful diolefinatation, 2) transesterification of the directing group with the solvent hexafluoroisopropanol (HFIP), and 3) moderate to good *meta*-selectivity. Some of these drawbacks probably originate from the conformational and electronic bias of arylacetic acid based substrates. We reasoned that a more electron-withdrawing SO₂ group may reduce electron density at the *ortho*- and *para*-position of the sulfone-based

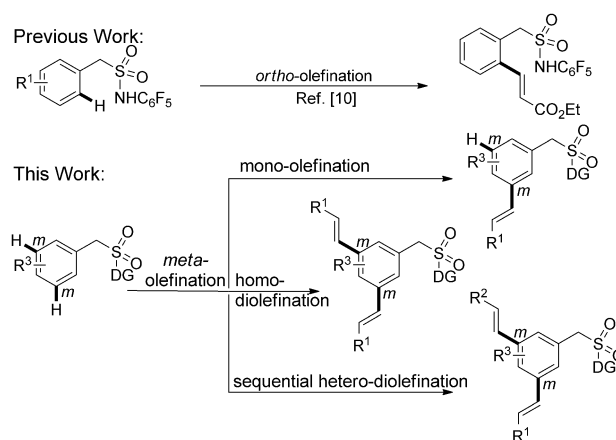
substrates, and that the flexible nature of the sulfonyl (SO₂), compared to carbonyl (CO) moiety, in the scaffold would enhance the directing power of the nitrile (Scheme 1).^[7] We



Scheme 1. Development of a new scaffold.

were particularly interested in the synthesis of di-alkenylated benzyldisulfonyl scaffolds because of the importance of diolefinated compounds in organic synthesis and materials chemistry.^[8] However the second olefination reaction is expected to be challenging since the electronic and steric properties of mono-olefinated species is completely different from those of the starting materials.^[9] Such challenges were overcome and we herein disclose an approach for *meta*-selective homo-diolefinatation and sequential hetero-diolefinatation of C–H bonds of sulfone-based arenes (Scheme 2). Although *ortho*-C–H olefination of benzyl sulfonamides^[10] and *ortho*-selective hetero-diolefinations have been documented,^[9] *meta*-selective hetero-diolefinatation is yet to be reported.

Initially, benzyldisulfonyl ester derivatives were tested with various of nitrile-based directing groups (Scheme 3; for structures see Figure 1).^[6,11] Upon optimization, 2-hydroxy-benzonitrile successfully delivered the diolefinatation product

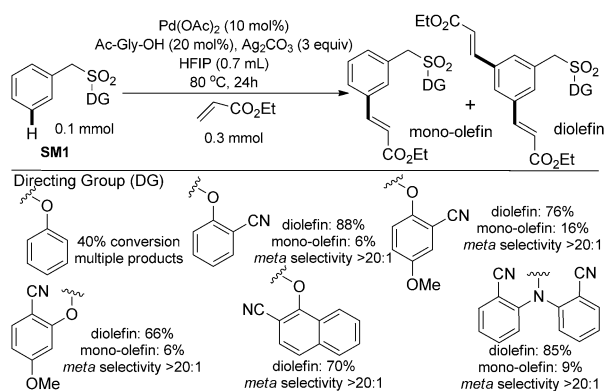


Scheme 2. Outline of this work.

[*] Dr. M. Bera, A. Maji, Dr. S. K. Sahoo, Prof. D. Maiti
Department of Chemistry, Indian Institute of Technology Bombay
Powai, Mumbai-400 076 (India)
E-mail: dmaiti@chem.iitb.ac.in

[**] This activity is supported by SERB, India (SB/S5/GC-05/2013).
Financial support was received from DST under the Fast Track
Scheme (M.B.), CSIR (A.M.), and IIT-B (S.S.).

Supporting information for this article is available on the WWW
under <http://dx.doi.org/10.1002/anie.201503112>.

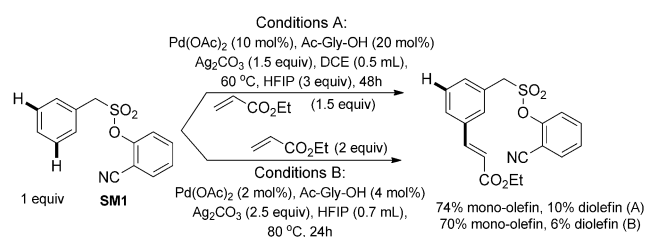


Scheme 3. Development of directing group. Yields and selectivities determined by ^1H NMR analysis of crude reaction mixture.



Figure 1. X-ray structures^[15] of SM1, 1f, and 4q.

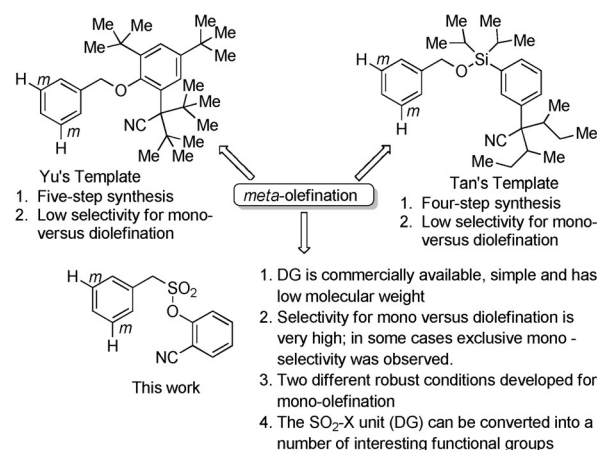
in 88% yield and with high selectivity (*meta*/other >20:1). Further optimization of the reaction parameters delivered two different catalytic conditions (A: with 10 mol% Pd in dichloroethane solvent and B: with 2 mol% Pd in HFIP solvent) for mono-olefination (Scheme 4).^[11] The major



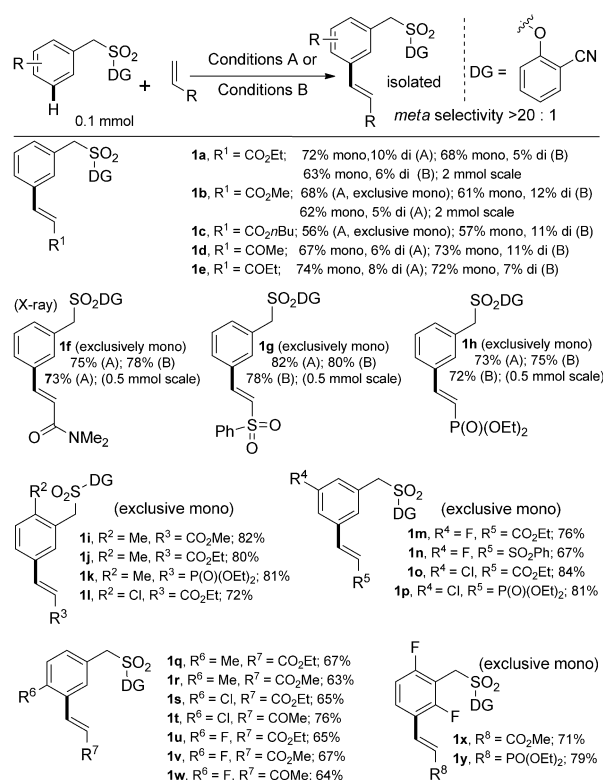
Scheme 4. Two protocols for mono-olefination. DCE = 1,2-dichloroethane. Yields and selectivities determined by ^1H NMR analysis of the crude reaction mixture.

advantages of the benzyisulfonate (our present scaffold) are as follows (Scheme 5): 1) the directing group is commercially available, simple, and has a low molecular weight. Notably, the templates of both Yu and Tan are synthesized through five- and four-step procedures, respectively; 2) in general, selectivity is very high (even for unbiased substrates). In a number of cases, exclusive mono-olefinated products were obtained. 3) two different robust conditions were developed for mono-olefination; 4) the $\text{SO}_2\text{-X}$ unit (DG) can be converted into a number of interesting groups (see Schemes 11 and 12).

Different acrylates and α,β -unsaturated ketones produced the desired mono-olefination products in moderate to good yields (1a–e; Scheme 6) with excellent *meta*-selectivity (*meta*/others >20:1). Olefins with amide, phosphonate, and sulfone moieties resulted in *meta*-selective mono-olefination products

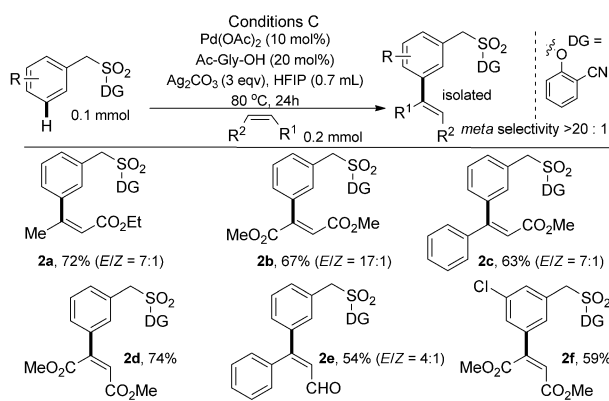


Scheme 5. Advantages of the present scaffold.



Scheme 6. The *meta*-selective mono-olefination. Yields and selectivities determined by ^1H NMR analysis of crude reaction mixture. See Figure 1 for X-ray structure.

exclusively (1f–h). Complete mono-selectivity was also obtained with both the *ortho*-methyl- and *ortho*-chloro-substituted benzyisulfonyl ester (1i–l) under the reaction conditions A. The *meta*-chloro- and fluoro-substituted benzyisulfonyl esters were olefinated selectively at the 5-position with different olefins (1m–p). In addition, this protocol is compatible with electron-donating and electron-withdrawing *para* substituents such as methyl (1q and 1r), chloro (1s and 1t), and fluoro (1u–1w). All *para*-substituted benzyisulfonyl esters produced 10–15% of the diolefinated product along with monoalkenylated products. The 2,6-difluoro sulfonyles-

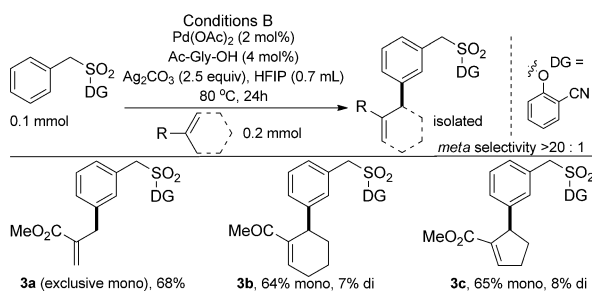


Scheme 7. Substrate scope with internal olefins. Yields and selectivities determined by ¹H NMR analysis of crude reaction mixture.

ter was olefinated to exclusively afford the *meta*-selective mono-olefinated products (**1x** and **1y**).

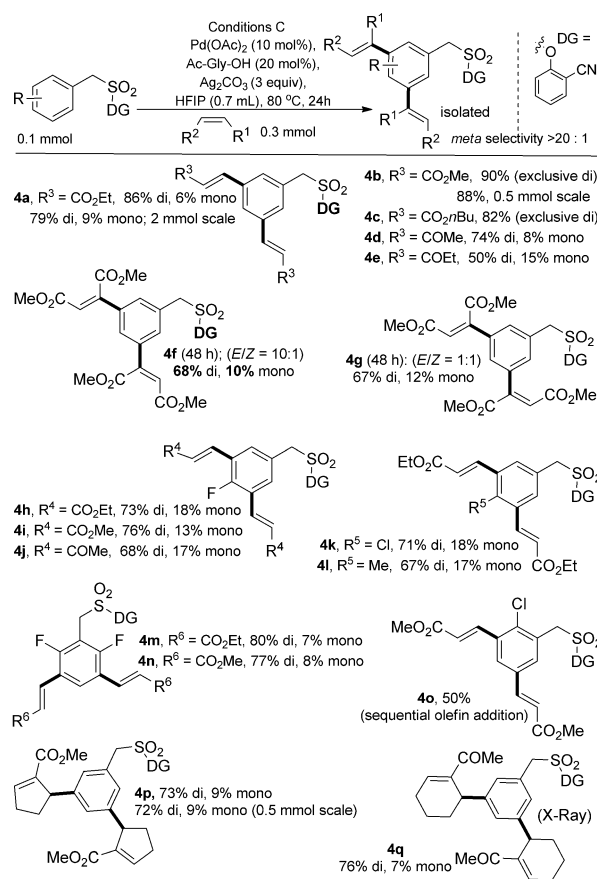
Interestingly under the reaction conditions C (Scheme 7), di-substituted *trans*-olefins (e.g. fumarate, crotonate, cinnamate, and cinnamaldehyde) were successfully employed.^[12] Notably, exclusive mono-selectivities were observed in all the entries. Although, a mixture of *E* and *Z* isomers were detected, the *E* isomer was generated as the major product (**2a–c** and **2e**). With a *cis*-olefin such as dimethyl maleate, *E* isomers were formed exclusively (**2d** and **2f**). These phenomena indicate that the stereoselectivity is thermodynamic in origin.^[13]

Similarly, reaction with methacrylate and cyclic trisubstituted olefins proceeded to provide *meta*-selective monoallyl benzene derivatives in good yields under the reaction conditions B (**3a–c**; Scheme 8). Formation of these allyl benzene derivatives could be explained by the regioselective β-H_b elimination rather than β-H_a elimination.^[11,14] Unfortunately this reaction is not compatible with electron-neutral or electron-rich olefins such as substituted styrenes.



Scheme 8. Formation of *meta*-allyl benzene. Yields and selectivities determined by ¹H NMR analysis of crude reaction mixture.

We subsequently explored homo-diolefination at the *meta* positions under the reaction conditions C (Scheme 9). Different acrylates, α,β-unsaturated ketones, and disubstituted internal olefins produced dialkenylated products in good yields (**4a–g**). Notably, chloro, fluoro, and methyl benzyldisulfonyl esters gave dialkenylated products (**4h–o**).



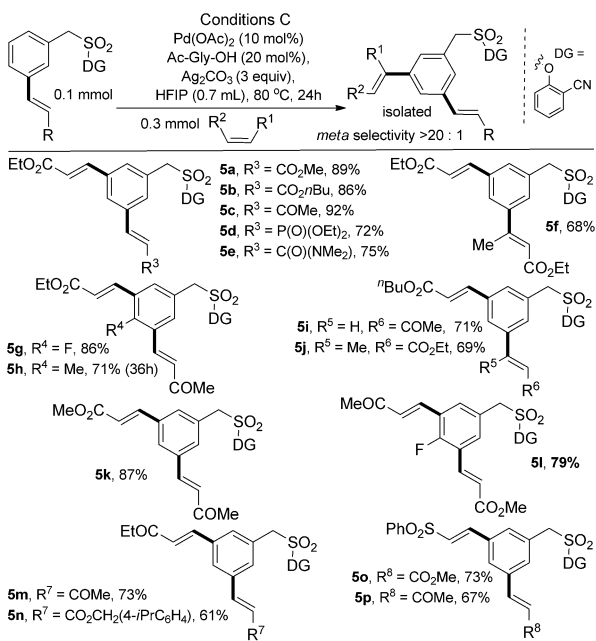
Scheme 9. Substrate scope for homo-diolefination. Yields and selectivities determined by ¹H NMR analysis of crude reaction mixture. See Figure 1 for X-ray structure.

The trisubstituted cyclic olefins also reacted well under the reaction conditions C to afford the *meta*-selective diallylbenzene derivatives in good yields (**4p** and **4q**).

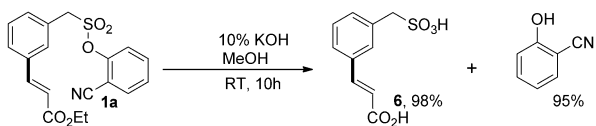
To obtain sequential olefination in a position-selective manner, mono-olefinated products were utilized. With the prefunctionalized olefinated product (Scheme 6), acrylates, methylvinyl ketone, ethylvinyl ketone, vinylsulfone, vinyl phosphonate, amide, and disubstituted olefins were incorporated at the remaining *meta*-position in an exclusive manner with good to excellent yields under the reaction conditions C (**5a–p**; Scheme 10). Hetero-diolefination reactions by sequential addition of two different olefins in a one-pot process gave the desired product (e.g., **5c**, 58% and **5k**, 52%) along with formation of other mono-olefinated and homo-diolefinated products.^[11]

The 2-hydroxybenzonitrile group is easily removed by hydrolysis at room temperature by using KOH in methanol to produce the *meta*-substituted diacid **6** in excellent yield (Scheme 11).

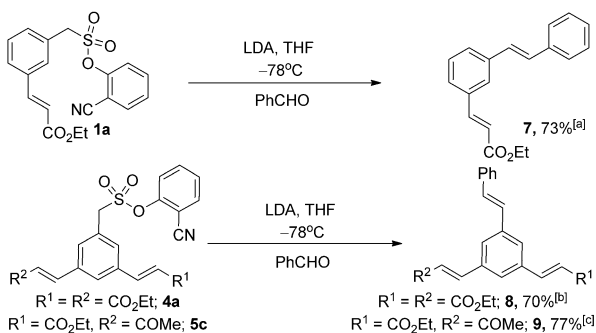
Additionally, the SO₂-DG scaffold can be easily cleaved by a modified Julia olefination. All mono-, di-, and hetero-diolefinated products, **1a**, **4a**, and **5c** were reacted with benzaldehyde in presence of LDA at –78 °C to afford the corresponding 1,3-divinyl and 1,3,5-trivinylbenzene derivatives, **7**, **8**, and **9** in good yields (Scheme 12).



Scheme 10. Sequential hetero-diolefination. Yields and selectivities determined by ^1H NMR analysis of crude reaction mixture.



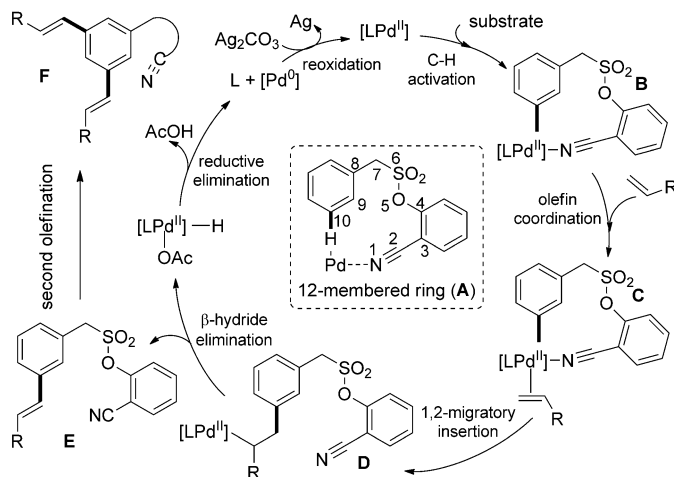
Scheme 11. Hydrolysis of directing group.



Scheme 12. Modified Julia olefination of **1a**, **4a**, and **5c**. [a] **7a** (Z isomer) 13%, **7b** (E isomer) 60%. [b] **8a** (Z isomer) 6%, **8b** (E isomer) 64%. [c] **9** (E isomer) 77%, (Z isomer) trace.^[11] LDA = lithium diisopropylamide, THF = tetrahydrofuran.

Kinetic experiments conducted on **SM1** to obtain the difunctionated product **4a** distinctly showed the intermediacy of the mono-olefinated product **1a**.^[11] A separate set of studies for the formation of **4a** from the mono-olefinated compound revealed a first-order rate dependency with respect to **1a**.^[11]

A proposal on the reaction course is presented in Scheme 13 on the basis of previous reports.^[4-6] The linear weakly binding nitrile group in the substrate is proposed to



Scheme 13. Proposed catalytic cycle.

make a 12-membered cyclophane-like pre-transition state (**A**). Because of the high electrophilicity, the nitrile-bound Pd^{II} center is likely to be reactive towards the *meta*-C–H bond. Formation of the mono-olefinated product **E** from **B** will involve 1) C–H activation, 2) olefin binding, 3) 1,2-migratory insertion, and 4) β-hydride elimination.

In summary, we have developed complementary methods for *meta*-selective mono-, di-, and sequential hetero-diolefinations of sulfone-based arene motifs with a 2-hydroxybenzonitrile template. Substituents are tolerated at all positions on the arene framework. The strategy for sequential *meta*-C–H olefination provides a novel route for the synthesis of *meta*-selective bis(alken)ylated compounds having two different olefins, compounds which are difficult to access using conventional methods.

Keywords: C–H activation · olefination · palladium · sulfur · synthetic methods

How to cite: *Angew. Chem. Int. Ed.* **2015**, 54, 8515–8519
Angew. Chem. **2015**, 127, 8635–8639

- [1] For selected examples on functionalization of arene C–H bonds, see: a) I. Moritani, Y. Fujiwara, *Tetrahedron Lett.* **1967**, 1119; b) C. Jia, T. Kitamura, Y. Fujiwara, *Acc. Chem. Res.* **2001**, *34*, 633; c) D. Alberico, M. E. Scott, M. Lautens, *Chem. Rev.* **2007**, *107*, 174; d) I. V. Seregin, V. Gevorgyan, *Chem. Soc. Rev.* **2007**, *36*, 1173; e) D.-H. Wang, K. M. Engle, B.-F. Shi, J.-Q. Yu, *Science* **2010**, *327*, 315; f) C.-L. Sun, B.-J. Li, Z.-J. Shi, *Chem. Rev.* **2011**, *111*, 1293; g) G. Rousseau, B. Breit, *Angew. Chem. Int. Ed.* **2011**, *50*, 2450; *Angew. Chem.* **2011**, *123*, 2498; h) T. Brückl, R. D. Baxter, Y. Ishihara, P. S. Baran, *Acc. Chem. Res.* **2012**, *45*, 826; i) S. R. Neufeldt, M. S. Sanford, *Acc. Chem. Res.* **2012**, *45*, 936; j) L. Ackermann, *Chem. Rev.* **2011**, *111*, 1315; k) J. Wencel-Delord, F. Glorius, *Nat. Chem.* **2013**, *5*, 369; l) J. Gallardo-Donaire, R. Martin, *J. Am. Chem. Soc.* **2013**, *135*, 9350; m) L. Zhou, W. Lu, *Chem. Eur. J.* **2014**, *20*, 634.
- [2] For selected reviews on metal-catalyzed directed C–H activation reactions, see: a) S. Murai, *Activation of Unreactive Bonds and Organic Synthesis*, Springer, Berlin, **1999**; b) G. Dyker, *Handbook of C–H Transformations*, Wiley-VCH, Weinheim, **2005**; c) R. Giri, B. F. Shi, K. M. Engle, N. Maugel, J. Q. Yu, *Chem. Soc.*

- Rev. **2009**, *38*, 3242; d) L. Ackermann, R. Vicente, A. R. Kapdi, *Angew. Chem. Int. Ed.* **2009**, *48*, 9792; *Angew. Chem.* **2009**, *121*, 9976; e) L. Ackermann, *Chem. Commun.* **2010**, *46*, 4866; f) I. A. I. Mkhaliid, J. H. Barnard, T. B. Marder, J. M. Murphy, J. F. Hartwig, *Chem. Rev.* **2010**, *110*, 890; g) C. S. Yeung, V. M. Dong, *Chem. Rev.* **2011**, *111*, 1215; h) C.-L. Sun, B.-J. Li, Z.-J. Shi, *Chem. Rev.* **2011**, *111*, 1293; i) L. McMurray, F. O'Hara, M. J. Gaunt, *Chem. Soc. Rev.* **2011**, *40*, 1885; j) J. Wencel-Delord, T. Droge, F. Liu, F. Glorius, *Chem. Soc. Rev.* **2011**, *40*, 4740; k) S. H. Cho, J. Y. Kim, J. Kwak, S. Chang, *Chem. Soc. Rev.* **2011**, *40*, 5068; l) P. B. Arockiam, C. Bruneau, P. H. Dixneuf, *Chem. Rev.* **2012**, *112*, 5879; m) G. Song, F. Wang, X. Li, *Chem. Soc. Rev.* **2012**, *41*, 3651; n) L. Louillat, F. W. Patureau, *Chem. Soc. Rev.* **2014**, *43*, 901; o) A. Ros, R. Fernandez, J. M. Lassaletta, *Chem. Soc. Rev.* **2014**, *43*, 3229; p) D. J. Musaev, T. M. Figg, A. L. Kaledin, *Chem. Soc. Rev.* **2014**, *43*, 5009; q) F. Zhang, D. R. Spring, *Chem. Soc. Rev.* **2014**, *43*, 6906; r) J. Schranck, A. Tlili, M. Beller, *Angew. Chem. Int. Ed.* **2014**, *53*, 9426; *Angew. Chem.* **2014**, *126*, 9580.
- [3] For *meta*-C–H functionalization using other approaches, see: a) J.-Y. Cho, M. K. Tse, D. Holmes, Jr., R. E. Maleczka, M. R. Smith III, *Science* **2002**, *295*, 305; b) T. Ishiyama, J. Takagi, K. Ishida, N. Miyaara, N. R. Anastasi, J. F. Hartwig, *J. Am. Chem. Soc.* **2002**, *124*, 390; c) R. J. Phipps, M. J. Gaunt, *Science* **2009**, *323*, 1593; d) Y.-H. Zhang, B.-F. Shi, J.-Q. Yu, *J. Am. Chem. Soc.* **2009**, *131*, 5072; e) O. Saidi, J. Marafie, A. E. W. Ledger, P. M. Liu, M. F. Mahon, G. Kociok-Kohn, M. K. Whittlesey, C. G. Frost, *J. Am. Chem. Soc.* **2011**, *133*, 19298; f) L. Zhou, W. Lu, *Organometallics* **2012**, *31*, 2124; g) N. Hofmann, L. Ackermann, *J. Am. Chem. Soc.* **2013**, *135*, 5877; h) C. Cheng, J. F. Hartwig, *Science* **2014**, *343*, 853; i) J. Luo, S. Preciado, I. Larrosa, *J. Am. Chem. Soc.* **2014**, *136*, 4109; j) A. J. Martínez-Martínez, A. R. Kennedy, R. E. Mulvey, C. T. O'Hara, *Science* **2014**, *346*, 834.
- [4] For palladium-catalyzed template-directed remote *meta*-C–H functionalization, see: a) D. Leow, G. Li, T.-S. Mei, J.-Q. Yu, *Nature* **2012**, *486*, 518; b) H.-X. Dai, G. Li, X.-G. Zhang, A. F. Stepan, J.-Q. Yu, *J. Am. Chem. Soc.* **2013**, *135*, 7567; c) R.-Y. Tang, G. Li, J.-Q. Yu, *Nature* **2014**, *507*, 215; d) G.-J. Cheng, Y.-F. Yang, P. Liu, P. Chen, T.-Y. Sun, G. Li, X. Zhang, K. N. Houk, J.-Q. Yu, Y.-D. Wu, *J. Am. Chem. Soc.* **2014**, *136*, 894; e) Y.-F. Yang, G.-J. Cheng, P. Liu, D. Leow, T.-Y. Sun, P. Chen, X. Zhang, J.-Q. Yu, Y.-D. Wu, K. N. Houk, *J. Am. Chem. Soc.* **2014**, *136*, 344; f) G. Yang, P. Lindovska, D. Zhu, J. Kim, P. Wang, R.-Y. Tang, M. Movassaghi, J.-Q. Yu, *J. Am. Chem. Soc.* **2014**, *136*, 10807; g) Y. Deng, J.-Q. Yu, *Angew. Chem. Int. Ed.* **2015**, *54*, 888; *Angew. Chem.* **2015**, *127*, 902.
- [5] Another template-directed *meta*-C–H functionalization, see: S. Lee, H. Lee, K. L. Tan, *J. Am. Chem. Soc.* **2013**, *135*, 18778.
- [6] For *meta*-C–H functionalization of phenylacetic acid by using cyanophenol derived ester, see: M. Bera, A. Modak, T. Patra, A. Maji, D. Maiti, *Org. Lett.* **2014**, *16*, 5760.
- [7] a) F. G. Bordwell, *Acc. Chem. Res.* **1988**, *21*, 456; b) I.-H. Um, S.-M. Chun, S.-K. Bae, *Bull. Korean Chem. Soc.* **2005**, *26*, 457; c) I. Chataigner, C. Panel, H. Gérard, S.-R. Piettre, *Chem. Commun.* **2007**, 3288.
- [8] a) J. H. Burroughes, D. D. C. Bradley, A. R. Brown, R. N. Marks, K. Mackay, *Nature* **1990**, *347*, 539; b) A. C. Grimsdale, K. L. Chan, R. E. Martin, P. G. Jokisz, A. B. Holmes, *Chem. Rev.* **2009**, *109*, 897.
- [9] For rare examples of sequential C–H activation, see: a) N. P. Grimster, C. Gauntlett, C. R. A. Godfrey, M. J. Gaunt, *Angew. Chem. Int. Ed.* **2005**, *44*, 3125; *Angew. Chem.* **2005**, *117*, 3185; b) A. García-Rubia, R. G. Arrayas, J. C. Carretero, *Angew. Chem. Int. Ed.* **2009**, *48*, 6511; *Angew. Chem.* **2009**, *121*, 6633; c) E. M. Beck, R. Hatley, M. J. Gaunt, *Angew. Chem. Int. Ed.* **2008**, *47*, 3004; *Angew. Chem.* **2008**, *120*, 3046; d) M. Nakano, H. Tsurugi, T. Satoh, M. Miura, *Org. Lett.* **2008**, *10*, 1851; e) S. Yanagisawa, K. Ueda, H. Sekizaawa, K. Itami, *J. Am. Chem. Soc.* **2009**, *131*, 14622; f) N. Umeda, K. Hirano, T. Satoh, M. Miura, *J. Org. Chem.* **2009**, *74*, 7094; g) K. M. Engle, D.-H. Wang, J.-Q. Yu, *Angew. Chem. Int. Ed.* **2010**, *49*, 6169; *Angew. Chem.* **2010**, *122*, 6305; h) A. Deb, S. Bag, R. Kancherla, D. Maiti, *J. Am. Chem. Soc.* **2014**, *136*, 13602; i) N. Schröder, F. Lied, F. Glorius, *J. Am. Chem. Soc.* **2015**, *137*, 1448.
- [10] H.-X. Dai, A. F. Stepan, M. S. Plummer, Y.-H. Zhang, J.-Q. Yu, *J. Am. Chem. Soc.* **2011**, *133*, 7222.
- [11] See the Supporting Information for detailed descriptions.
- [12] a) M. D. K. Boele, G. P. F. Van Strijdonck, A. H. M. De Vries, P. C. J. Kamer, J. G. De Vries, P. W. N. M. Van Leeuwen, *J. Am. Chem. Soc.* **2002**, *124*, 1586; b) G. Cai, Y. Fu, Y. Li, X. Wan, Z. Shi, *J. Am. Chem. Soc.* **2007**, *129*, 7666.
- [13] a) C. Gürtler, S. L. Buchwald, *Chem. Eur. J.* **1999**, *5*, 3107; b) P. Wucher, L. Caporaso, P. Roesle, F. Ragone, L. Cavallo, S. Mecking, I. Göttker-Schnetmann, *Proc. Natl. Acad. Sci. USA* **2011**, *108*, 8955; c) S.-R. Ban, H.-N. Wang, V. Toader, D. S. Bohle, C.-J. Li, *Org. Lett.* **2014**, *16*, 6282.
- [14] a) G. T. Crisp, *Chem. Soc. Rev.* **1998**, *27*, 427; b) A. F. Littke, G. C. Fu, *J. Am. Chem. Soc.* **2001**, *123*, 6989; c) V. Caló, A. Nacci, A. Monopoli, A. Detomaso, P. Iliade, *Organometallics* **2003**, *22*, 4193; d) P. Fristrup, S. L. Quement, D. Tanner, P.-O. Norrby, *Organometallics* **2004**, *23*, 6160; e) E. W. Werner, M. S. Sigman, *J. Am. Chem. Soc.* **2010**, *132*, 13981.
- [15] CCDC 1043571 (**SM1**), 1043569 (**1f**), and 1043570 (**4q**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [16] P. R. Blakemore, *J. Chem. Soc. Perkin Trans. 1* **2002**, 2563.

Received: April 5, 2015
Revised: April 29, 2015
Published online: June 9, 2015